

## Observations and Modeling of the Green Ocean Amazon (GOAMAZON): Particulate Matter and Gases Final Campaign Summary

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# Observations and Modeling of the Green Ocean Amazon (GOAMAZON): Particulate Matter and Gases Final Campaign Summary

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## **Executive Summary**

Because of their proven adverse effects on human health and vegetation, and also considering their influence over the local and regional climate, inhalable fine particles (PM<sub>2.5</sub>) and NO<sub>2</sub>, SO<sub>2</sub>, and O<sub>3</sub> have been collected at the ARM site located in Manacapuru, Amazon, Brazil, as a part of the GoAmazon 2014/5 project. PM<sub>2.5</sub> samples were analyzed through gravimetry, black carbon transmittance, elemental composition by energy dispersive x-ray fluorescence, and ionic concentration (cations) by ion chromatography. NO<sub>2</sub> and SO<sub>2</sub> samples were analyzed by ion chromatography, whereas O<sub>3</sub> samples were analyzed through ultraviolet-vis spectrophotometry. Sampling of both particulate and gaseous pollutants took place during the two intensive operation periods (IOP1 from February to March 2014, and IOP2 from August to October 2014). Results are interpreted both separately and as a whole with the specific goal of identifying compounds that could affect the population's health and/or could act as cloud condensation nuclei. Chemical analysis supports the elucidation of the possible origins, transport mechanisms, health effects, and main effects of the assessed pollutants in those environments.

## **Acronyms and Abbreviations**

ARM Atmospheric Radiation Measurement (program)

GoAmazon 2014/5 Green Ocean Amazon 2014//2015 project

IOP1 First Intensive Operation Period, February to March 2014IOP2 Second Intensive Operation Period, August to October 2014

NO<sub>2</sub> Nitrogen dioxide

O<sub>3</sub> Ozone

PM<sub>2.5</sub> Particulate matter of aerodynamic diameter equal to or smaller than 2.5 μm

SO<sub>2</sub> Sulfur dioxide

WHO World Health Organization

XRF X-ray fluorescence

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## 1.0 Background

Industrial and urban development; growth of the automotive fleet; current patterns of consumption; deforestation, agriculture, and fertilizer use; and recurrent forest burning are some of the factors that have contributed to the increase of atmospheric pollutant emissions. The scientific literature agrees that this air pollution is a significant threat to the health of the world population, especially in developing countries (WHO 2006, Bruce et al. 2000, Pope et al. 2004, Urdea et al. 2006, Freitas et al., 2013; Watson, 2014; Kessler, 2014; Fowler et al., 2013), where regulations and inspections in general are less stringent. In recent decades, a strong correlation indicates the association between a high level of air pollution and its effects on human health, such as mortality rates, hospitalization for respiratory problems and cardiovascular diseases, worsening of asthma attacks, adverse effects on lung function, low weight of newborns, high infant and perinatal mortality, tuberculosis, cataract, and nasopharyngeal, larynx, and lung cancer (Bruce et al., 2000; Pope et al., 2004, Pope et al. 2002, Delfino et al. 2003, Valavanidis et al. 2008, Watson 2014, Kessler, 2014).

All of the main sources of atmospheric pollutants—industrial activity, vehicular exhaust, and biomass burning (forest fires and deforestation)—are present in the Amazon region of Brazil. Prevailing northwest winds carry a large portion of the pollution arising from Manaus, the capital city of the state of Amazonas, to the ARM sampling site in Manacapuru, which is located in a rural area nearly 70 km southwest of Manaus,. Manacapuru has no well-developed urban center; thus, local anthropogenic sources of air pollution can be excluded from measurements taken at the ARM site. Therefore, this site allows the assessment of anthropogenic influences in a place where urban development is minimal and pollutants that are present are mainly transported by wind.

For this campaign, both particulate and gaseous pollutants were assessed. For particulate pollution, the finest fraction of particulate matter (i.e., particulates with an aerodynamic diameter equal to or smaller than 2.5  $\mu$ m, also called PM<sub>2.5</sub>) is the most harmful because it is able to penetrate the alveoli and therefore reach the bloodstream. For that reason, the PM<sub>2.5</sub> fraction was selected for sampling. For gaseous pollution, the inorganic gases nitrogen and sulfur dioxides (NO<sub>2</sub> and SO<sub>2</sub>, respectively) and ozone (O<sub>3</sub>) were selected because of their proven harmful effects on human health and vegetation. There also is evidence that these pollutants influence climate processes such as cloud formation for PM<sub>2.5</sub>, atmospheric oxidation potential (NO<sub>2</sub> and O<sub>3</sub>), and rainfall acidity (NO<sub>2</sub> and SO<sub>2</sub>).

Particulate and gaseous samples were taken at the ARM site in Manacapuru, Brazil, as a part of the GoAmazon 2014/5 project, which is commonly called T3. Gaseous samples were taken weekly during both the wet season from February 1, 2014, to March 31, 2014 (during the first Intensive Operation Period, or IOP1) and the dry season from August 15, 2014 to October 15, 2014 (during the second Intensive Operation Period, or IOP2). Sampling was carried out using passive diffusive samplers from Radiello® at five points (hereby called T3 01, T3 02, T3 03, T3 04 and T3 05) distributed along a 5 km straight line perpendicular to the Manacapuru-Manaus axis (the point in the middle, T3 03, was located inside the Manacapuru ARM site, and two points were located at each side from the center). Particulate sampling was taken daily also during the wet season (March 3, 2014, to April 22, 2014) and the dry season (September 4, 2014 to October 18, 2014), but at only one point at the site (T3 03). PM<sub>2.5</sub> samples were taken by a Harvard impactor using 37-mm diameter polycarbonate filters from Nucleopore®.

All samples were analyzed in the Federal University of Paraná, Brazil. NO<sub>2</sub> and SO<sub>2</sub> samples were analyzed by ion chromatography (using an ion chromatograph Dionex Model ICS-5000 from Thermo Fisher, analytic column IonPac® AS19, eluent potassium hydroxide with concentration gradient, and 0.3 mL min⁻¹ flux with 100-μL loop), and O₃ samples were analyzed using ultraviolet-vis spectrophotometry. PM₂.₅ filters were analyzed by gravimetry (using an analytic microscale Sartorius with 0.1-μg resolution), black carbon content was analyzed by transmittance (using an optic transmisometer SootScan Model OT 21 from Magee Scientific Company), soluble cation concentration was analyzed using an analytic column (IonPac® CS12, eluent methanesulfonic acid with isocratic concentration, 0.3-mL min⁻¹ flux with 100-μl loop), and elemental composition was analyzed by X-ray fluorescence (XRF) (using a Minipal 4 from Panalytical). Filters from IOP1 were analyzed by all four techniques mentioned, while IOP2 filters were analyzed only for black carbon content and soluble cations concentration.

This work was financially supported by the National Council for Scientific and Technological Development, with additional support from the Fundação de Amparo à Pesquisa do Estado do Amazonas and the Financiadora de Estudos e Projetos. We also acknowledge the support from the Central Office of the Large Scale Biosphere Atmosphere Experiment in Amazônia, the Instituto Nacional de Pesquisas da Amazônia, and the Universidade do Estado do Amazonas. We want to give special recognition and express our appreciation to Bruno Takeshi and Professor Rodrigo Souza and his team for their extraordinary help during these campaigns.

We thank the Atmospheric Radiation Measurement (ARM) Climate Research Facility and the U.S. Department of Energy Office of Science user facility sponsored by the Office of Biological and Environmental Research.

## 2.0 Notable Events or Highlights

During the two sampling periods, we did not register any instrument issues, and we did not record unusual meteorological episodes.

#### 3.0 Lessons Learned

In our initial plans for this project, we intended to analyze gaseous pollutants other than  $NO_2$ ,  $SO_2$ , and  $O_3$ , namely acetic and formic acid, dioxins, and the mono-aromatic compounds benzene, toluene, ethylbenzene and meta-, orto- and para-xylenes. Because of logistical complications, sampling of dioxins was not possible, and was excluded from the project. The mono-aromatic compounds and organic acids were sampled at the site, but the samples are still being analyzed so those results are not included in this report.

Regarding particulate matter analysis, we initially intended to perform single-particle sampling, in addition to PM<sub>2.5</sub> sampling, and to perform additional analyses, namely single-particle elemental and molecular compositions by micro Raman spectrometry, and electron probe micro-analysis using facilities for low-Z element determination. These analyses and sampling of single particles were not performed because of unforeseen issues regarding the necessary equipment.

## 4.0 Results

The results obtained by the present project are divided into the following sections, the first one dealing with the gaseous pollutants and the second with particulate matter.

#### 4.1 Gaseous Pollutants

Table 1, Table 2, and Table 3 present the results obtained for NO<sub>2</sub>, SO<sub>2</sub>, and O<sub>3</sub>, respectively, for all five sampling points. Further, Figure 1 shows the average NO<sub>2</sub>, SO<sub>2</sub>, and O<sub>3</sub> concentrations graphically for each site superimposed onto a satellite image of the studied area.

**Table 1.** NO<sub>2</sub> concentrations measured at the five Manacapuru site sampling points.

				NO <sub>2</sub> (μg m <sup>-3</sup> )		
	Sampling Periods	T01	T02	T03	T04	T05
	3/1-8/2014 <sup>a</sup>	23.317 ±7.1323	0.100 ±0.1815	$\mathrm{BDL}^{\mathrm{b}}$	12.199 ±0.7388	17.96 ±6.6433
IOD1	3/8-15/2014	$8.935 \pm 0.23$	$0.162 \pm 0.0759$	BDL	$2.709 \pm 0.3424$	$9.257 \pm 0.1287$
IOP1	3/15-24,/2014	$7.411 \pm 0.429$	$1.72 \pm 0.0495$	BDL	$2.764 \pm 0.1055$	$0.124 \pm 0.0373$
	3/24-31/2014	$5.213 \pm 0.9308$	$2.17 \pm 0.3527$	BDL	$1.853 \pm 0.0604$	$1.186 \pm 0.0108$
	8/27-9/3/2014	$0.827 \pm 0.0237$	2.121 ±1.4439	4.446 ±0.1872	6.132 ±0.4739	$1.737 \pm 0.1023$
	9/3-10/2014	BDL	$1.512 \pm 0.0667$	$1.605 \pm 0.0718$	$1.137 \pm 0.442$	$0.514 \pm 0.341$
LOD2	9/10-17/2014	BDL	$1.336 \pm 0.5082$	3.413 ±0.2997	$2.129 \pm 0.0082$	$0.782 \pm 0.2625$
IOP2	9/17-10/1/2014	c		$3.114 \pm 0.113$	$2.336 \pm 0.3611$	$1.801 \pm 0.1109$
	10-1-8/2014		$1.737 \pm 3.0825$	1.109 ±1.6299	$2.021 \pm 0.0244$	$1.477 \pm 0.5939$
	10/8-15/2014		$1.319 \pm 0.0476$	$2.959 \pm 0.5036$	$0.339 \pm 0.0935$	$2.631 \pm 0.576$

<sup>&</sup>lt;sup>a</sup> Date convention is month/day/year.

<sup>&</sup>lt;sup>b</sup> Below Detection Limit

<sup>&</sup>lt;sup>c</sup> Missing samples

Table 2. SO<sub>2</sub> concentrations measured at the five Manacapuru site sampling points.

				$SO_2 (\mu g m^{-3})$		
	Sampling period	T01	T02	T03	T04	T05
IOP1	3/1-8/2014 <sup>a</sup>	0.276 ±0.0737	0.137 ±0.0156	$0.079 \pm 0.0002$	0.316 ±0.0028	0.436 ±0.0949
	3/8-15/2014	$0.177 \pm 0.0098$	$0.156 \pm 0.01$	$0.152 \pm 0.0049$	$0.256 \pm 0.0164$	3.675 ±0.0181
	3/15-24,/2014	$0.197 \pm 0.0071$	$0.253 \pm 0.0022$	$0.166 \pm 0.0212$	$0.202 \pm 0.0183$	$1.078 \pm 0.0084$
	3/24-31/2014	$0.584 \pm 0.036$	$0.264 \pm 0.0019$	$0.196 \pm 0.0074$	$0.122 \pm 0.0014$	$0.622 \pm 0.0154$
IOP2	8/27-9/3/2014	4.505 ±0.035	1.584 ±0.4433	8.963 ±0.1406	8.125 ±0.2396	14.88 ±0.3131
	9/3-10/2014	$0.143 \pm 0.005$	$1.679 \pm 0.0056$	$1.259 \pm 0.037$	$1.17 \pm 0.1995$	$0.272 \pm 0.0375$
	9/10-17/2014	b	$0.47 \pm 0.0756$	$0.815 \pm 0.0512$	$1.652 \pm 0.0106$	$0.284 \pm 0.0367$
	9/17-10/1/2014			$0.209 \pm 0.0056$	$0.303 \pm 0.0017$	$0.207 \pm 0.0189$
	10-1-8/2014		$0.215 \pm 0.1519$	$0.157 \pm 0.0626$	$22533 \pm 0.0426$	$0.216 \pm 0.0376$
	10/8-15/2014		$0.323 \pm 0.0459$	$0.453 \pm 0.0589$	1,001 ±0.0107	$0.33 \pm 0.0357$

<sup>&</sup>lt;sup>a</sup> Date convention is month/day/year.

**Table 3.** O<sub>3</sub> concentrations measured at the five Manacapuru site sampling points.

				O <sub>3</sub> (μg m <sup>-3</sup>	)	
	Sampling period	T01	T02	T03	T04	T05
	3/1-8/2014 <sup>a</sup>	3.3265 <sup>b</sup>	5.8898	1.5969	4.545	6.2578
IOD1	3/8-15/2014	8.6066	7.3148	1.8168	4.0788	13.5495
IOP1	3/15-24./2014	11.3441	5.4738	1.439	3.6268	11.9792
	3/24-31/2014	10.8555	6.5128	3.8097	3.1291	5.1396
	8/27-9/3/2014	6.7162	1.8587	1.4296	4.553	3.2996
	9/3-10/2014	c	1.3384	1.0003	-	6.2148
LOD2	9/10-17/2014		1.6535	2.1597	18.4487	12.7573
IOP2	9/17-10/1/2014		1.7151	2.3724	40.213	36.0312
	10-1-8/2014		1.2657	1.3327	16.3307	19.4015
	10/8-15/2014		1.2584	1.557	1.5665	22.7877

<sup>&</sup>lt;sup>a</sup> Date convention is month/day/year.

<sup>&</sup>lt;sup>b</sup>Missing samples

 $<sup>^{\</sup>rm b}$  Standard deviation was not calculated for  ${\rm O_3}$  samples because the volume extracted during the analytical procedure was not enough to analyze replicates of each sample.

<sup>&</sup>lt;sup>c</sup> Missing samples

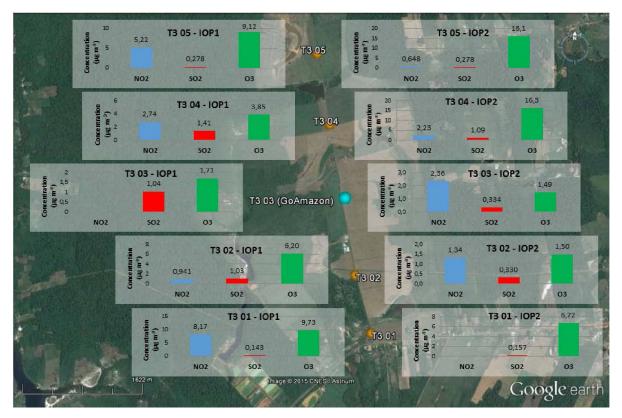


Figure 1. Satellite image of the studied area with superimposed average concentrations of NO<sub>2</sub>, SO<sub>2</sub>, and O<sub>3</sub> for each sampling point during each sampling period.

#### 4.2 Particulate matter

The results from PM<sub>2.5</sub> analyses are provided in the following sections for sampling point T3 03 during both IOPs.

#### 4.2.1 Weight Concentration

Figure 2 displays all the weight concentration data resulting from measurements taken at sampling point T3 03 during IOP1.

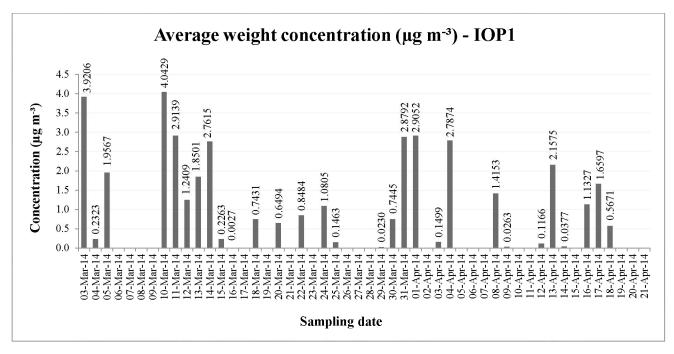


Figure 2. Daily average weight concentration measured at T3 03 during IOP1.

#### 4.2.2 Black Carbon Content

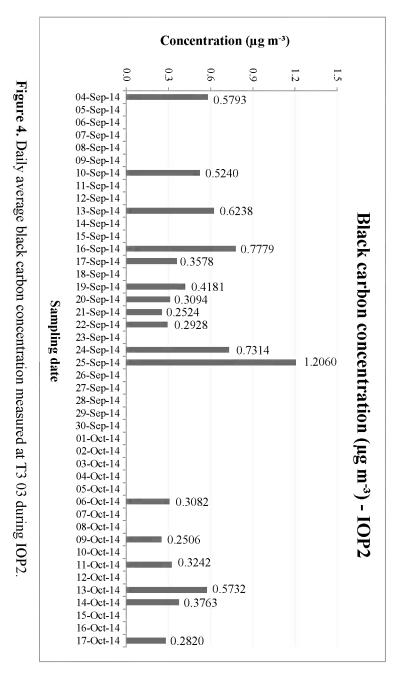
Figure 3 and Figure 4 display all the black carbon concentration data resulting from measurements taken at sampling point T3 03 during IOP1 and IOP2, respectively.

#### 4.2.3 Elemental Composition by X-Ray Fluorescence

Figure 5 shows the content of silicon (Si), sulfur (S), potassium (K), iron (Fe), calcium (Ca) and titanium (Ti) present in the PM<sub>2.5</sub> filters collected at sampling location T3 03 during IOP1 (measured by means of XRF).

#### 4.2.4 Soluble Cation Concentrations

Figure 6 and Figure 7 show the content of the soluble cations lithium (Li<sup>+</sup>), sodium (Na<sup>+</sup>), ammonium (NH<sub>4</sub><sup>+</sup>), potassium (K<sup>+</sup>), magnesium (Mg<sup>2+</sup>), and calcium (Ca<sup>2+</sup>) in the PM<sub>2.5</sub> filters collected at sampling location T3 03 during IOP1 and IOP2, respectively (measured by ion chromatography).



Concentration (µg m<sup>-3</sup>) 03-Mar-14 0.6695 =0.321304-Mar-14 05-Mar-14 0.3213 06-Mar-14 07-Mar-14 08-Mar-14 09-Mar-14 10-Mar-14 = 0.653811-Mar-14 0.3478 Black carbon 12-Mar-14 = 0.414713-Mar-14 0.2788 14-Mar-14 = 0.451415-Mar-14 = 0.445516-Mar-14 = 0.348917-Mar-14 18-Mar-14 0.782819-Mar-14 20-Mar-14 0.1069 concentration (µg m<sup>-3</sup>) - IOP1 21-Mar-14 22-Mar-14 = 0.680923-Mar-14 Sampling 24-Mar-14 0.1002 25-Mar-14 0.2787 26-Mar-14 27-Mar-14 28-Mar-14 29-Mar-14 0.2916 30-Mar-14 0.2134 31-Mar-14 0.2814 01-Apr-14 0.0762 02-Apr-14 03-Apr-14 0.2933 04-Apr-14 -0.498705-Apr-14 06-Apr-14 07-Apr-14 08-Apr-14 = 0.290009-Apr-14 -0.336010-Apr-14 11-Apr-14 12-Apr-14 0.4697 13-Apr-14 ■ = 0.419214-Apr-14 0.1655 15-Apr-14 16-Apr-14 17-Apr-14 0.0944 18-Apr-14 19-Apr-14 20-Apr-14 21-Apr-14 0.1769

Figure 3. Daily average black carbon concentration measured at T3

03 during IOP1

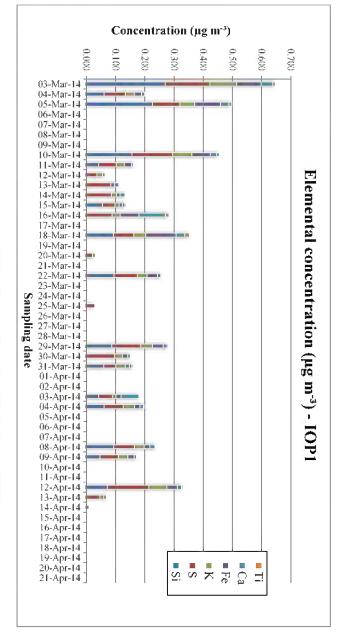


Figure 5. Elemental concentration of the PM<sub>2.5</sub> filters collected at T3 (XRF measurements). 03 during IOP1

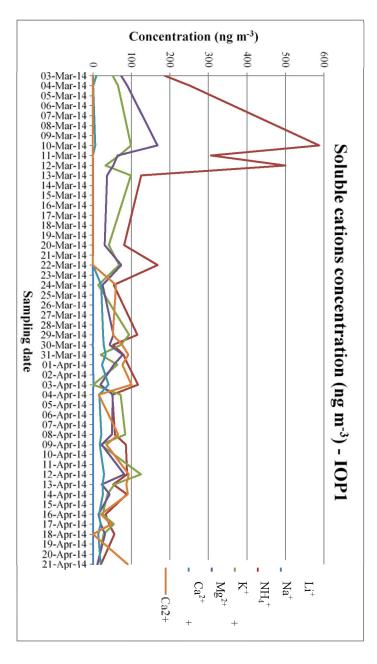
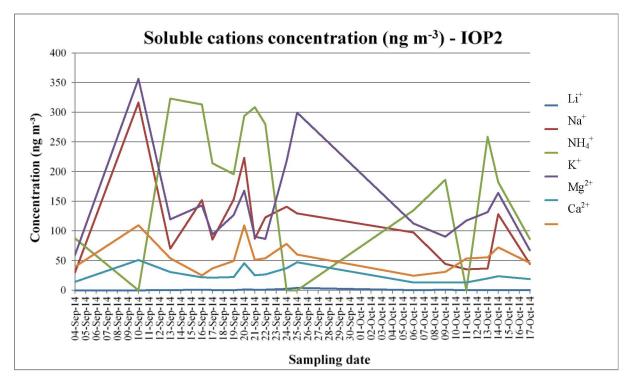


Figure 6. Soluble cation concentrations of the PM<sub>2.5</sub> filters collected at T3 (ion chromatography measurements). 03 during IOP



**Figure 7.** Soluble cation concentrations of the PM<sub>2.5</sub> filters collected at T3 03 during IOP2 (ion chromatography measurements).

## 5.0 Public Outreach

Not applicable.

## 6.0 GoAmazon 2014/5 Publications

## 6.1 Journal Articles/Manuscripts

Not applicable.

## 6.2 Meeting Abstracts/Presentations/Posters

The results presented here were also presented as a poster during the GoAmazon2014/5 Science Conference in Cambridge, MA, 18-20 May 2015. The poster presented there is reproduced below:

#### MONITORING AIR POLLUTANTS AT T1 MANAUS AND T3 MANACAPURU DURING GOAMAZON2014/5 IOPS

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#### SUMMARY

In the present study, 4 additional sites were selected for sampling nearby the T3 site, forming a straight line with 5 sampling points, being the original T3 in the center. This strategy was pursuit in order to provide an unsprecedented spatial gradient of concentration of several air pollutants, namely NO<sub>2</sub>, SO<sub>2</sub>, O<sub>3</sub>, H<sub>2</sub>S, PM<sub>5,3</sub> and BC.

#### SAMPLING SITES

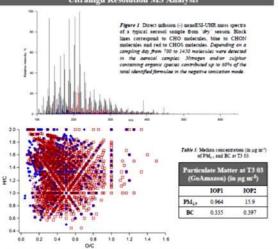
Sampling was carried out at five sites near T3 (the GoAmazon experiment) during the Intense Operation Periods 1 and 2 (IOP1 and 2. The location of the five sites, along with their coordinates, is shown in Figure 1, whereas Table 1 details the periods when the samples were taken.

Samples of PM2.5 were taken by means of a Harvard impactor, and analyzed for gravumetry and BC content. Gaseous pollutants were sampled using passive diffusive samplers, provided by Radiello®, as shown in Figure 2, which were later analyzed by ion which were later analyzed by ion chromatography (NO<sub>2</sub> and SO<sub>2</sub>) or UV-spectrophotometry (O<sub>2</sub> and H<sub>2</sub>S). Compounds under analyze: BTEX Acetic Acid and Formic Acid.

A state-of-the-art direct-infusion electrospray (nanoESI) ultrahigh resolution mass spectrometry was used to determine hundreds of influenced by biogenic-anthropogenic interactions. Comprehensive mass spectral data evaluation methods (e.g. Kendrick Mass Defect and Van Krevelen dagram) were used to identify compound clauses and mass distributions of the compound classe detected species.

Table 1. Sampling period of	particul	ate parte
Particulate daily	samplin	
Period		
03 Mer 2014 to 31 Mer 2014 (18 samples)	1071	Wet
04 Sep 3014 to 14 Oct 3014 (16 saugsles)	10P2	Dry

Table 2. Sampling period of	gaseous	compoun
Gazes weekly sa	aplies	
Period	-	
01 Mar 2014 to 06 Mar 2014 08 Mar 2014 to 15 Mar 2014 15 Mar 2014 to 24 Mar 2014 24 Mar 2014 to 31 Mar 2014	10P1	Wet Season
27 Aug 2014 to 63 Sep 2014 63 Sep 2014 to 10 Sep 2014 16 Sep 2014 to 17 Sep 2014 17 Sep 2014 to 61 Oct 2014 61 Oct 2014 to 68 Oct 2014	10P2	Dry Season



0.0 0.2 0.4 0.6 0.8 10 1.2 1.4 1.6 OC
Figure J. Vas Screeken diagrams showing CRO molecules from "ser" (thise diamonds) and 'dry' (red squares) seasons people from the 'dry' (seed occurred more entitled molecules (with depths of C rams) compared to that phon the 'dry suggesting that we deplections (specificates) offers molecules compared on the specific specifi

## MEDIAN CONCENTRATION OF GASEOUS POLLUTANTS AT T3 IOP 1 IOP , (GoAmazon) **●** O<sub>3</sub> Google ear Figure J. Median concentration (in µg m²) of various gaseous compounds at T3 and 2.5 km for from there (scales differ from dataset to dataset) A JAPESP ACNPG

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